

# CBCS SCHEME

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17NT35

## Third Semester B.E. Degree Examination, Dec.2018/Jan.2019 Physical and Chemical Principles of Nano Technology

Time: 3 hrs.

Max. Marks: 80

**Note: Answer any FIVE full questions, choosing ONE full question from each module.**

### Module-1

- 1 a. Derive an expression for uncertainty principle with the help of localization experiment. (10 Marks)  
b. Write a note on Molecular Orbital theory and its applications. (10 Marks)

OR

- 2 a. Write a note on Valance Bond theory and its applications. (10 Marks)  
b. State and explain Debroglie hypothesis. (05 Marks)  
c. State and explain Planck's hypothesis. (05 Marks)

### Module-2

- 3 a. Define Thermodynamics. Write a note on importance and limitations of thermodynamics. (10 Marks)  
b. Write a note on Enthalpy and Heat capacity of a system. (10 Marks)

OR

- 4 a. State and explain the first law of thermodynamics with mathematical expressions. (10 Marks)  
b. Explain the concept of heat capacity at constant volume and constant pressure. (06 Marks)  
c. What is Spontaneous process? Write the criteria for spontaneity. (04 Marks)

### Module-3

- 5 a. Explain Debye theory of molar heat capacity. Write the limitations. (10 Marks)  
b. Distinguish between Metal, Insulator and Semiconductor. (06 Marks)  
c. Explain Brillouin zone. (04 Marks)

OR

- 6 a. Explain Einstein's theory of molar heat capacity. Write the limitations. (10 Marks)  
b. Discuss the concept of lattice vibrations and lattice waves in a linear mono atomic lattice. (10 Marks)

### Module-4

- 7 a. Explain the classification of semiconductors with examples. (10 Marks)  
b. Explain the construction and working of Tunnel diode with suitable graph. Write uses of Tunnel diode. (10 Marks)

OR

- 8 a. Explain P – N junction semiconductor diode and give its advantages. (10 Marks)  
b. Write a note on Rectification. (10 Marks)

Important Note : 1. On completing your answers, compulsorily draw diagonal cross lines on the remaining blank pages.  
2. Any revealing of identification, appeal to evaluator and /or equations written eg, 42+8 = 50, will be treated as malpractice.

Module-5

- 9 a. Explain the dynamic properties of colloids. (10 Marks)  
b. Explain the characteristics and identification of type of emulsions. (05 Marks)  
c. Explain Crystalloids and Colloids. (05 Marks)

OR

- 10 a. Explain classification of colloids based on state of aggregation of the dispersed phase and dispersed medium with example. (10 Marks)  
b. Explain Electrophoresis and Electro osmosis. (08 Marks)  
c. Explain Tyndall effect. (02 Marks)

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